Name: $\qquad$
Date: $\qquad$

1. What is the density of a rock that has a mass of 234 g and a volume of $7.9 \mathrm{~cm}^{3}$ ?

$$
26.9 \mathrm{~g} / \mathrm{cm}^{3}
$$

2. A piece of metal that has a density of $5.2 \mathrm{~g} / \mathrm{cm}^{3}$ and a mass of 100 g was placed in a full jar of water. How many mL of water spilled out of the jar? Note: one $\mathrm{cm}^{3}=$ one mL .
19.2 mL
3. A huge meteor had a mass of $3.2 \times 10^{12} \mathrm{~g}$. If the density of the meteor was $4.2 \mathrm{~g} / \mathrm{cm}^{3}$, what was the volume of the meteor?

$$
7.6 \times 10^{11} \mathrm{~cm}^{3}
$$

4. Gold was mined in California and divided into 1.5 kg pieces. Each piece of gold was a perfect cube 8.1 cm tall. What was the density of the gold?

$$
0.0028 \mathrm{~kg} / \mathrm{cm}^{3}
$$

5. A very large boulder with a volume of 1200 L has a mass of $1.4 \times 10^{9} \mathrm{~g}$. What is the density of the boulder?

1,200,000 g/L
6. What is the density of a piece of wood that has a mass of 2.74 g and a volume of $3.10 \mathrm{~cm}^{3}$ ?
$0.884 \mathrm{~g} / \mathrm{cm}^{3}$
7. Find the volume of a liquid if 32.5 g of the liquid has a density of $0.852 \mathrm{~g} / \mathrm{mL}$.
38.1 mL

